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Massarilactones A and B: novel secondary metabolites from the freshwater aquatic fungus *Massarina tunicata*

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Abstract—Massarilactones A and B (1 and 2) have been isolated from cultures of the freshwater aquatic fungus *Massarina tunicata*. The structures, including absolute stereochemistry, were determined by X-ray diffraction analysis of their bis(4-bro-mobenzoate) derivatives. © 2001 Elsevier Science Ltd. All rights reserved.

Our preliminary studies of freshwater aquatic fungi have led to the isolation of several new bioactive compounds,¹⁻³ including three sesquiterpenoid metabolites described recently from the aquatic fungus *Massarina tunicata* Shearer & Fallah (A-25-1; = ATCC 201760).¹ Investigations of scale-up cultures of *M. tunicata* have led to the isolation of two new polyketide-derived antibacterial lactones that we have named massarilactones A and B (1–2), both of which contain unusual ring systems. Details of the isolation and structure elucidation of **1** and **2** are presented here. Fractionation of the ethyl acetate extract of *M. tunicata* liquid cultures by chromatography on silica gel, followed by Sephadex LH-20, and/or reversed-phase HPLC, afforded compounds **1** and **2**.⁴ The molecular formula of massarilactone A (**1**) was determined to be $C_{11}H_{14}O_5$ (five unsaturations) on the basis of NMR and HRFABMS data. The ¹H and ¹³C NMR data (Table 1) and DEPT results for massarilactone A suggested the presence of an ester group, a -CHCH₃ moiety, an *sp*³ methylene unit, four oxymethine protons, and an oxygenated 1,1-disubstituted double bond. These data ac-



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C#	$^{1}\mathrm{H}^{\mathrm{a}}$ δ (mult; $J_{\mathrm{H-H}}$ in Hz)	$^{13}\mathrm{C^b}~\delta$
1	_	171.7
3	_	156.6
4	5.69 (br t; 2.1)	76.9
5	_	61.5
6	1.97 (br q; 7.2)	46.2
7	4.22 (m)	83.6
8	2.26 (dd; 16, 3.3)	38.6
	1.82 (ddd; 16, 5.7, 1.8)	
9	4.26 (br dd; 5.4, 5.4)	66.4°
10	4.22 (m)	69.6°
11	4.79 (br t; 2.7)	89.7
	4.58 (br dd; 2.7, 2.1)	
12	1.15 (d; 7.2)	14.3

^a Recorded at 300 MHz.

^b Recorded at 75 MHz.

^c Assignments may be interchanged.

counted for all but two exchangeable protons, and indicted that the structure of massarilactone A (1) is tricyclic. ${}^{1}H{-}{}^{1}H$ decoupling experiments identified a -CH₂CHO- subunit, and revealed that the *exo*-methylene signals (H₂-11) were allylically coupled to the oxymethine proton signal at δ 5.69 (H-4). Treatment of massarilactone A (1) with acetic anhydride resulted in the formation of a diacetate (3), allowing assignment of both exchangeable protons as secondary alcohol groups. The structure of massarilactone A was assigned as 1 on the basis of detailed NMR analysis and was ultimately confirmed by X-ray diffraction analysis.

The bis(4-bromobenzoate) ester (4) of massarilactone A was prepared by treatment of 1 with 4-bromobenzoyl chloride. Crystals of 4 suitable for analysis by X-ray crystallography were obtained by slow evaporation of an acetone solution. The final X-ray crystallographic model of 4 (Fig. 1)⁵ revealed the structure and absolute stereochemistry of massarilactone A, as shown in 1.



C#	$^{1}\mathrm{H^{a}}~\delta$ (mult, $J_{\mathrm{H-H}}$ in Hz)	¹³ C ^b δ	Selective INEPT ^a correlations
1		172.4	
3	4.81 (q, 6.9)	74.2	1, 4, 9, 10
4		177.8	
6	4.67 (br dd, 8.2, 6.6)	84.3	4, 7, 8, 11, 12
7	3.80 (dd, 5.1, 6.6)	71.5	8, 9, 11
8	4.49 (br d, 5.1)	64.2	1, 4, 6, 7, 9
9		100.3	
10	1.43 (d, 6.9)	17.1	3, 4
11	5.66 (ddq, 15, 8.1,	124.9	
	1.5)		
12	5.90 (ddq, 15, 6.6,	133.5	6, 11, 13
	0.7)		
13	1.73 (dd, 6.6, 1.5)	17.8	

^a Recorded at 300 MHz.

^b Recorded at 75 MHz.

NMR assignments for 1 were made on the basis of chemical shifts, DEPT data, and comparison of its spectral data with those from the known compound spirostaphylotrichin F (5).⁶

Massarilactone B (2)⁴ was determined to be an isomer of 1 on the basis of HRFABMS and ¹³C NMR data, but these data also suggested the presence of significant structural differences. The ¹H, ¹³C, and DEPT NMR data (Table 2) for 2 indicated the presence of two methyl groups, four oxymethine units, a *trans*-disubstituted olefin, and two exchangeable protons. Decoupling experiments permitted the assignment of two isolated spin systems corresponding to an isolated -OCHCH₃ moiety (C3–C10) and a trioxygenated *trans*-2-hexene unit. The chemical shifts of the three remaining, nonprotonated carbons (δ 177.8, 172.4, 100.3) suggested that they comprise a β-alkoxy-α,β-unsaturated lactone unit.⁷ The two exchangeable protons were assigned to hydroxy groups at C-7 and C-8 after analysis of ¹H



Figure 1. Final X-ray model of 4.



Figure 2. Final X-ray model of 7.

NMR data for the diacetate 6 formed by treatment of 2 with acetic anhydride. The remainder of the structure and NMR assignments for 2 were proposed on the basis of selective INEPT data (Table 2). As was the case for 1, X-ray crystallographic analysis⁵ of the bis-(4-bromobenzoate) ester of 2 (7; Fig. 2) confirmed the structure of massarilactone B and permitted assignment of its absolute stereochemistry as shown.

To our knowledge, the methanofuro[3,4-b]oxepin ring system found in **1** has not been previously described, although metabolites that contain the similar methanooxepino[2,3-c]pyrrole ring system (e.g. spriostaphylotrichin F; **5**) have been reported from *Staphylotrichum coccosporum*.⁶ Similarly, it appears that no natural products having the furo[3,4-b]pyran ring system found in massarilactone B (**2**) have been previously reported. However, larger ring systems incorporating such a system are known, and a synthetic intermediate possessing this ring system has been prepared.⁸

Massarilactones A and B both appear to be derived from the same type of polyketide precursor, with addition of a three-carbon unit accounting for carbons 3, 4, and 11. These compounds bear close biogenetic resemblance to several other fungal metabolites, including rosigenin, the curvupallides, and the spirostaphylotrichins.^{9–11} Biosynthetic studies of members of this class (e.g. **5**) have suggested that they are formed by condensation of a polyketide chain with an unidentified C_4 unit, most likely either an amino acid (e.g. aspartic acid) or a citric acid cycle intermediate.^{10,11}

Massarilactones A (1) and B (2) exhibited antibacterial activity against *Bacillus subtilis* (ATCC 6051) in standard disk assays, affording zones of inhibition of 19 and 16 mm, respectively, at 200 µg/disk. Massarilactone B was also active against *Staphylococcus aureus* (ATCC 29213) at the same level, causing a zone of inhibition of 12 mm. Neither compound showed activity in assays against *Aspergillus flavus* (NRRL 6541), *Fusarium verticillioides* (ATCC 24378), or *Candida albicans* (ATCC 14053) at 200 µg/disk.

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References

- 1. Oh, H.; Shearer, C. A.; Gloer, J. B. J. Nat. Prod. 1999, 62, 497–501.
- Harrigan, G. G.; Armentrout, B. L.; Shearer, C. A.; Gloer, J. B. J. Nat. Prod. 1995, 58, 1467–1469.
- Xu, X.; DeGuzman, F. S.; Shearer, C. A.; Gloer, J. B. J. Org. Chem. 1992, 57, 6700–6703.
- 4. Massarilactone A (1): 34 mg obtained from 8 L of fermentation broth; $[\alpha]_D + 8.7^{\circ}C$ (*c* 0.3 g/dL; 24°C; CH₂Cl₂); UV (CH₃OH) 210 (ε 2200); ¹H and ¹³C NMR data, Table 1; HRFABMS (NaI/3-NBA) obsd *m/z* 249.0720, calcd for C₁₁H₁₄O₅+Na, 249.0739. Massarilactone B (2): 70 mg from 8 L of broth; $[\alpha]_D -109^{\circ}$ (*c* 2.2 g/dL; 28°C; CH₃OH); UV (CH₃OH) 236 (ε 9000), 270 (ε 6300); ¹H, ¹³C, and selective INEPT NMR data, Table 2; HRFABMS (LiI/glycerol) obsd *m/z* 227.0912, calcd for C₁₁H₁₄O₅+H, 227.0919.
- 5. X-Ray data for 4 were collected on an Enraf-Nonius CAD4 diffractometer (Mo K α radiation) using θ -2 θ scans. The structure was solved using a MULTAN direct methods program, and refined using full-matrix leastsquares. Crystals of 4 (0.42×0.20×0.08 mm) were monoclinic (space group $P2_1$) with cell dimensions a=9.443(3), b = 17.089(5), c = 7.552(3) Å. The 7822 measurements vielded 4147 independent reflections (309 parameters) after equivalent data were averaged and Lorenz and polarization corrections were applied. The final refinement gave $R_1 = 0.0561$, $wR_2 = 0.0878$. X-Ray analysis of compound 7 was performed at the University of Minnesota using a Siemens SMART system at 173(2) K. Crystals of 7 (0.45×0.19×0.045 mm) were also monoclinic (space group $P2_1$) with cell dimensions a = 7.69990(10), b = 28.0555(3), c = 11.6064(2) Å. The specimen was determined to be a rotational twin with the twin law (transposed, by rows) [1, 0, -1/7; 0, -1; 0, 0, -1]. The 8489 measurements yielded 8489 independent reflections (619 parameters). The final refinement gave $R_1 = 0.0914$ and $wR_2 = 0.2088$. Atomic coordinates for both compounds have been deposited at the Cambridge Crystallographic Data Centre.
- 6. Sandmeier, P.; Tamm, C. Helv. Chim. Acta 1989, 72, 784–792.
- Sohár, P. Nuclear Magnetic Resonance Spectroscopy; CRC Press: Boca Raton, Florida, 1983; Vol. I, pp. 67–68 and references cited therein.
- Paquette, L. A.; Sivik, M. R. Synth. Commun. 1991, 21, 467–479.
- Renaud, J.-M.; Tsoupras, G.; Tabacchi, R. Helv. Chim. Acta 1989, 72, 929–932.
- 10. Ayer, W. A.; Craw, P. A. J. Can. J. Chem. 1992, 70, 1348–1355.
- 11. Sandmeier, P.; Tamm, C. Helv. Chim. Acta 1989, 72, 774–783.